ANTIMUTAGENIC ACTIVITY OF SESBANIA JAVANICA MIQ. FLOWER DMSO EXTRACT AND ITS MAJOR FLAVONOID GLYCOSIDE

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Abstract. The antimutagenic activity of *Sesbania javanica* Miq. or Sano, an edible vegetable flower DMSO extract against aflatoxin B1 (AFB₁) benzo (a) pyrene [B(a)P], was evaluated by means of the Ames' test. The *Sesbania javanica* Miq. flower DMSO extract showed a strong inhibitory effect against AFB₁ and B(a)P mutagens. A search to isolate the major flavonoid in *Sesbania javanica* Miq. flower extract found the flavonol glycoside, Quercetin $3-2^{G}$ -rhamnosylrutinoside, which was confirmed by its physicochemical properties as a major constituent of the flower. Quercetin $3-2^{G}$ -rhamnosylrutinoside (207 µg/plate) also showed a strong inhibitory effect against AFB₁ and B(a)P with a more than 70% inhibition rate.

INTRODUCTION

Research suggests that a number of dietary components may be capable of increasing or decreasing cancer incidence (Wattenberg, 1983). Thus, manipulation of the diet may be a noninvasive approach to minimize cancer incidence (Liu et al, 1992). Increased consumption of fruits and vegetables has been found to be associated with a lower incidence of various types of cancer and lower cancer mortality rates in several human cohort and case-control studies for all common cancer sites (Ames et al, 1993; Willett, 1994). Intake of sufficient amounts of antimutagens and/or anticarcinogens is believed to confer a preventive effect against the development of human cancer (Ferguson, 1994). In animal experiments, vegetables common in human diets have been shown to have antimutagenic effects (Bingham, 1990; Wattenberg and Coccia, 1991; Vinitketkumnuen et al, 1994; Kusamran et al, 1998). Fruit and vegetables provide protection against mutagenicity and cyto-

Tel: 66(0) 5526-1000 Ext 6236; 01-7402181; 09-7026988; Fax: 66 (0) 5526-1935 E-mail: surapon14t@yahoo.com toxicity which has been attributed to the various antioxidants contained in them (Ames, 1983; Gey, 1990; Cozzi *et al*, 1997). There is evidence indicating that free radicals cause oxidative damage to proteins, lipids, and nucleic acids. Therefore, antioxidants, which can neutralize free radicals, may be of importance in the prevention of several disease states. Fruits and vegetables contain a great variety of different antioxidant compounds. Most of the antioxidant capacity of fruit and vegetables comes from compounds other than vitamins C, E, β -carotene and flavonoid (Wang *et al*, 1996; Pietta, 2000; Yang *et al*, 2001).

Vegetables, such as Cha-om, Indian mulberry, cucumber, ivy gourd, lettuce, neem, sweet basil, lemon grass, sadao, and krachia daeng act as antimutagens against the mutagenicity of benzo (a) pyrene [B(a)P] and aflatoxin B₁ (AFB₁) (Viniketkumnuen *et al*, 1994; Kusamran *et al*, 1998; Nakahara *et al*, 2002). A wide variety of naturally occurring compounds or their mixtures, have been shown to inhibit mutagenicity and carcinogenicity cased by several chemicals. Humans can be exposed to AFB₁ or B(a)P by either ingestion or inhalation. AFB₁ is common in cultures and food products. Many agricultural products are vulnerable to attack by fungi (*Aspergillus flavus, Aspergillus parasiticus*, etc)

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which are able to produce toxic metabolites called mycotoxins. Among the various mycotoxins, AFB1 has assumed significance due to its deleterious effects on humans, poultry and livestock. AFB, is primarily a hepatotoxin, and in some species, such as the rat and the rainbow trout, a hepatocarcinogen and mutagen. B(a)P is one of a group of compounds called polycyclic aromatic hydrocarbons (PAHs) which are found in the exhaust of motor vehicles and other gasoline and diesel engines, as well as in the emissions of coal-, oil-, and wood-burning stoves and furnaces, cigarette smoke, general soot and smoke of industrial, municipal, and domestic origin, and in cooked foods, especially charcoal-broiled foods, in incinerators, coke ovens, and in asphalt processing and use. It may be found in some foodstuffs and in some public and private drinking water supplies. It is a carcinogen, mutagen, tumorigen, neoplastigen and teratogen. Toxic response depends on how the molecule is metabolized, both mutagens require metabolic activation to DNA-reactive intermediates by cytochrome P450-mediated mixed-function oxidases (MFOs) in order to exert their carcinogenic action. Both these mutagens tested were chosen because they are the most frequently occurring in mutagens found in food and feed.

Sesbania javanica Miq. flower or "Sano" in Thai or "Phak hong hang" (northern Thai). The Sesbania javanica Miq. flower is a leguminous plant (Leguminosae) originally grown in Southeast Asia, especially in Thailand. The Sano (Sesbania javanica Miq.) flowers are harvested only in the rainy season. It is commonly consumed in Thailand. It has also been used for its diverse medicinal properties. In ancient Thai traditional medicine, it was used as an anti-inflammatory for insect bites, detoxicification, intestinal abscess healing, stomach discomfort and to relieve internal fever and thirst.

Flavonoids are a diverse group of phytochemicals that are produced by various plants in high quantities. Natural flavonoids usually occur as glycosides (*eg* glucose, rhamnose, rhamnoglucosides, and rutinosides). Flavonoids have anti-bacterial, anti-viral, anti-inflammatory, antiangional, analgesic, anti-allergic hepatoprotective, cytostatic, apoptotic, estrogenic, anti-estrogenic (Hodek *et al*, 2002), and antioxidant properties (Pietta, 2000). Their ability to scavenge hydroxyl redicals, superoxide anion radicals and lipid peroxyradicals highlights many of their flavonoid helth-promoting functions, which are important in the prevention of diseases associated with oxidative damage of membranes, proteins and DNA (Pietta, 2000; Ferguson, 2001) and the ability to modulate (inhibit or stimulate) several enzymes or cell receptors (Zhai *et al.* 1998; Omiecinski *et al*, 1999; Ciolino and Yeh, 1999).

In our study of the antimutagenic activity of DMSO extract of Sano (*Sesbania javanica* Miq), commonly used in the local diet, was observed. The major flavonoid constituent of the Sano (*Sesbania javanica* Miq.) flower was analyzed using a UV-diode array, ESI-MS, and ¹H, ¹³C NMR. The quantitative values of this major flavonoid were determined.

MATERIALS AND METHODS

Plant material extraction

The Sesbania javanica Miq. flower was purchased in Bangkok. The petals were frozen at -70°C and lyophilized. Freeze-dried petals were blended to powder. Ground, freeze-dried samples were extracted with DMSO and then centrifuged at 10,000*q* for 10 minutes to remove the flower debris and filtered through a 0.25 μ m sterile membrane filter disc before being subjected it to antimutagenicity testing. The other 0.5 g ground freeze-dried sample was extracted with 10 ml of MeOH/DMSO (1:1) (Kawaii et al, 2000). The sample was mixed and centrifuged at 10,000g for 10 minutes to collect the extract solvent. These steps were repeated 3 times. The extracts were combined and MeOH was added to give a total volume of 20 ml.

Mutagens

AFB₁ and B(a)P were purchased from the Sigma Chemical (St Louis, USA). These mutagens were dissolved in DMSO. The AFB₁ concentration was 0.3 μ g/plate for the *Salmonella* tester strains TA 98 and TA 100. The B(a)P concentrations were 10 μ g/plate for the *Salmonella* tester strain TA 98 and 5.0 μ g/plate for the *Salmonella* tester strain tester strain te

potent mutagens and carcinogen agents. Precautions were taken for the proper handling and disposal of these chemicals.

Bacterial test strain

The Salmonella typhimurium TA 98 (frameshift) and TA 100 (base-pair substitution) strains which were histidine dependent due to a mutation of the histidine operon, were kindly provided by the research division of the National Cancer Institute, Bangkok, Thailand, and maintained as described by Maron and Ames (1983). The genotypes of the test strains were checked routinely for their histidine requirement.

S9 fraction and S9 mix

The method described by Matsushima et al (1976) was followed. Male Sprague-Dawley rats (from The Animal Center of Salaya, Mahidol University) were injected intraperitoneally with phenobarbital and β -naphthoflavone. Four days after injection, the rats were sacrificed, their livers were removed and minced in 0.15 M KCl, and then homogenized using a Potter-Elvehjem apparatus. The homogenate was centrifuged at 9,000g for 20 minutes in a refrigerated centrifuge, and the supernatant S9 fraction was decanted and immediately stored in cryogenic vials (Wheaton) of 1-2 ml, at -70°C. The protein concentration of the S9 fraction was determined by the Lowry method (Lowry et al, 1951). The protein concentration required approximate 30 mg/ml. The S9 required for the preparation of the S9 mix was thawed at room temperature and placed in a container of crushed ice. The S9 mix was prepared as soon as the S9 had thawed. The components of the S9 mix were sodium phosphate-KCI buffer 13.2 ml (0.5M sodium phosphate, pH 7.4 15.15 ml, 1M KCl 2.5 ml added to DW 32.35 ml), 0.16M MgCl₂ 1.0 ml, 0.1M glucose-6-phosphate 1.0 ml, 0.1M NADP 0.8 ml, and S9 4.0 ml, mixed and kept in the ice box. The S9 mix was prepared freshly for each mutagenicity assay.

Antimutagenicity test

A modified plate incorporation procedure (Maron and Ames, 1983) was employed to determine the effect of the *Sesbania javanica* Miq. flower DMSO extracts on AFB_1 and B(a)P induced mutagenicity. In this study, *Sesbania*

javanica Mig. flower DMSO extract was tested at 3 doses: 0.05, 0.5 and 5.0 mg/plate (100 μl). The major isolated compound was tested at 207 μg/plate in DMSO (100 μl). AFB₁ (0.3 μg/plate for TA 98 and TA 100) and B(a)P (10 µg/plate for TA 98; 5 µg/plate for TA 100) were used as standard mutagens. In brief, mutagen (0.1 ml) was distributed in sterilized capped tubes, and then 0.1 ml of tested S. typhimurium bacterial suspension from an overnight culture (1 x10⁹ to 2 x 10⁹ cells/ml) and 0.1 ml of Sesbania javanica Mig. flower DMSO extracts were added. Then, 0.5 ml of S9 mix was added. The entire mixture was pre-incubated (Matsushima et al, 1980), while shaking, at 37°C for 20 minutes before 2 ml of molten top agar was added, then the mixture was poured onto a minimal medium agar plate. The plates were incubated at 37°C for 48 hours, then the His⁺ revertant colonies on each plate were counted. Each sample was assayed using duplicate plates. The data is presented as the mean ± standard error of the two independent assays. Plates without mutagens and without Sesbania javanica Miq. flower DMSO extracts were considered as negative controls and plates with mutagens as positive controls. The antimutagenicity effect is expressed as percentage of inhibition (% inhibition) = 100 - $(R_1/R_0 x)$ 100), where R_1 is the number of His⁺ revertant/ plate of plates exposed to mutagens and Sesbania javanica Miq. flower extracts and the R_{o} is the number of His⁺ revertant/plate of the positive controls. The number of spontaneous revertants was substracted from the numerator and denominator. The mutagenicity of the mutagens (positive control) in the absence of Sesbania javanica Mig. flower DMSO extract was defined as 0% of the inhibition. The antimutagenic effect was considered moderate when the inhibitory effect of the Sesbania javanica Miq. flower DMSO extract was in the range of 25-40%, and strong when the inhibitory effect was >40%. An inhibitory effect <25% was considered weak, and was not recognized as a positive result.

Analytical HPLC was conducted on a SpectraSystem (P4000, AS3000, UV3000) liquid chromatography fitted with a C-18 ODS Hypersil (Thermo, Electron corporation, UK), reverse phase 4.6 x 250 mm (5 μ m) column. HPLC analysis of the flavonoids was performed primarily according to the method of Owen *et al* (2000; 2003); 20 μ l of *Sesbania javanica* Miq. flower extract was injected. Detection of flavonoids was carried out by means of a UV-detector set at an absorbance of 270 nm at room temperature. The mobile phase consisted of 2% acetic acid in water (B) and methanol (A) utilizing the following gradient over a total run time of 45 minutes: 95% B for 2 minutes, 75% B in 8 minutes, 60% B in 20 minutes, 50% B for 10 minutes, and 0% B until completion of the run. The flow rate of the mobile phase was 1 ml/minute.

Semi-preparative HPLC was conducted on the same SpectraSystem liquid chromatography fitted with a reverse-phase C-18 ODS Hypersil (Thermo, Electron corporation, UK), 25 cm (10 μ m) column (internal diameter, 10 mm) similar to that used for analytical HPLC. The flow rate of the mobile phase was 3.0 ml/minute, with a 50 μ l injection volume. Peaks eluting from the column were collected and pooled, then the solvent was removed by evaporation and lyophilization.

Acid hydrolysis

From each fraction collected by HPLC, 50 μ l of the methanolic suspension were dried under a stream of nitrogen and 0.5 M H₂SO₄ in double distilled water (100 μ l) was added and vertexed. The solutions were incubated at 100°C for 1 hour. After incubation, double distilled water (1.0 ml) was added, passed through a Sep-Pak Vac 1-cc C-18 cartridge (Waters). The water elute (800 μ l) was collected and identified by paper chromatography (Markham, 1982) using different sugars as standards.

Analytikjena-specord S100 spectrophotometer

The UV absorption data of these compounds in methanol were recorded on the analytikjena-specord S100 (Jena, Germany), by photo diode array was performed at a wavelength of 200-550 nm at the Department of Chemistry, Faculty of Science, Naresuan University.

Electrospray Ionization-Mass Spectrometry (ESI-MS)

The ESI mass spectra were recorded on a Bruker Esquire spectrometer (Bruker-Franzen

Analytik) in methanol in the positive mode at the service division of the Department of Chemistry, Faculty of Science, Mahidol University.

Nuclear Magnetic Resonance Spectroscopy (NMR)

The NMR spectra were recorded on a Bruker Avance 400 NMR spectrometer (Bruker Biospin AG) at the Department of Chemistry, Faculty of Science, Naresuan University. The ¹H and ¹³C nuclear magnetic resonance spectra were recorded at 400.13 MHz and 100 MHz, respectively, at 25°C using the standard technique. The compounds (ca. 4-8 mg) were dissolved in 0.4 ml methanol- d_{4} (99.96%D) in 5 mm sample tubes. The following measurements were carried out: conventional 1D ¹H and ¹³C. The chemical shifts (d) were reported in ppm relative to the solvent resonances. Methanol- d_4 was used as an internal secondary standard at $\delta_{_{\rm H}}$ 3.31 ppm for ¹H. The coupling constant (J) values were recorded in Hz; the $\delta_{\rm C}$ = 49.05 ppm for ¹³C.

Molecular modeling

As a guide for making the initial molecular formula and the NMR signal assignments, chemical shift predictions were made using the ChemDraw Ultra 6.0 (CambridgeSoft, Cambridge, MA).

RESULTS

The antimutagenicity test, the DMSO extract of Sesbania javanica Miq. flower strongly inhibited AFB1 and B(a)P induced mutagenicity, by S. typhimurium tester strains TA98 and TA100 (Table 1). The relationship between the percentage of inhibition versus the extract concentration was analyzed by simple regression. Analyses were carried out using Microsoft Excel. The inhibitory effect of Sesbania javanica Miq. flower extract on AFB₁ (r=0.894; p≤0.01) and B(a)P (r=0.813; p≤0.01) for Salmonella TA 98 was obtained. The inhibitory effect of the Sesbania javanica Mig. flower extract on AFB, (r=0.816; $p \le 0.01$) and B(a)P (r=0.803; $p \le 0.01$) for Salmonella TA 100, increased with an increasing concentration of flower extract. The isolated major flavonoid (207 µg/plate) also possessed a strong antimutagenic effect (>70%) against AFB1 and

$(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ 0.05 278 ± 9^a 58.7^b 148 ± 9 60.4 350 ± 9 60.2 266 ± 6	
(mg) AFB_1 (%) $B(a)P$ (%) AFB_1 (%) $B(a)P$ $0.3 \ \mu g$ inhibition $10 \ \mu g$ inhibition $0.3 \ \mu g$ inhibition $5 \ \mu g$ in $(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ $(+S9)$ 0.05 278 ± 9^a 58.7^b 148 ± 9 60.4 350 ± 9 60.2 266 ± 6	
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$0.5 238 \pm 6 65.3 127 \pm 4 67.4 288 \pm 4 69.9 223 \pm 7$	71.6
5.0 201 ± 9 71.5 116 ± 10 71.1 255 ± 10 75 202 ± 10	76.2
Positive control 630 ± 9 328 ± 12 738 ± 12 - 548 ± 10	-
Spontaneous 30 ± 2 30 ± 2 94 ± 9 - 94 ± 9	-

Table 1The antimutagenicity of Sesbania javanica Miq. flower DMSO extract on S. typhimurium TA98and TA 100.

The significance was tested by Student's t- test (p<0.01) vs control group.

 $^{\rm a}$ Values represent mean \pm SD based on two experiments.

 $^{\rm b}$ The values in parenthesis are the inhibition rates (%).

Inhibition (%inhibition) = 100 - ($R_1/R_0 \ge 100$), where

 R_1 is the number of His⁺ revertant/plate of plates exposed to mutagens and *Sesbania javanica* Miq. flower extract R_0 is the number of His⁺ revertant /plate in the positive control.

The number of spontaneous revertants was substracted from the numerator and denominator.

Table 2

Antimutagenic effects of the major flavonol, Quercetin 3-2^G-rhamnosylrutinoside from the *Sesbania javanica* Miq. flower DMSO/MeOH (1:1) extract against AFB₁ and B(a)P on *S. typhimurium* TA98 and TA 100.

Quercetin 3 - 2 ^G - rhamnosyl rutinoside (µg/plate)	Number of His ⁺ revertants/plate of <i>S. typhimurium</i> TA98			Number of His ⁺ revertants/plate of <i>S. typhimurium</i> TA100				
	AFB ₁ 0.3 μg/plate (+S9)	(%) inhibition	B(a)P 10 μg/plate (+S9)	(%) inhibition	AFB ₁ 0.3 μg/plate (+S9)	(%) inhibition	B(a)P 10 μg/plate (+S9)	(%) inhibition
207 Positive control spontaneous	166±9 ^a 596±10 31±2	76.1 ^b	112±8 369±8 31±2	76.0	280±10 824±14 87±5	73.8	252±9 729±10 87±5	74.3

The significance was tested by Student *t*- test (p < 0.01) vs control group.

^aValues represent mean ± SD based on two experiments.

^bThe values in parenthesis are the inhibition rates (%).

Inhibition (%inhibition) = 100 - ($R_1/R_0 \times 100$), where

 R_1 is the number of His⁺ revertant/plate of plates exposed to mutagens and *Sesbania javanica* Miq. flower extract R_0 is the number of His⁺ revertant /plate in the positive control.

The number of spontaneous revertants was substracted from the numerator and denominator.

B(a)P (Table 2).

In a search for the major flavonoid, the DMSO/MeOH (1:1) extract was reverse-phase high performance chromatographed over C-18

ODS Hypersil to yield the major compounds. The major compound was a flavonol glycoside isolated from the *Sesbania javanica* Miq. flower DMSO/MeOH: a pale-yellow amorphous powder. On acid hydrolysis of this compound, glucose, and rhamnose were identified by paper chromatography (Markham, 1982). UV spectra of this compound in MeOH showed a $\lambda_{\rm max}$ of 357.81 and 256.67 nm as quercetin.

The physicochemical properties of this major compound were as follows: ESI-MS in the positive mode, showed a molecular ion of compound 1 at *m/z* 779.2 (M + Na)⁺, with fragment ions 691.3, 413.2, and 301.1, which correspond to the formula C₃₃H₄₀O₂₀ m/z 756.2 (M+H)+; λ_{max} in MeOH of 357.81 and 256.67 nm.¹H NMR chemical shifts (CD₃OD) of aglycone were δ 6.18 (*d*, *J*=2.2 Hz, H-6), 6.38 (*d*, J= 2.2 Hz, H-8), 7.58 (d, J = 2.1 Hz, H-2'), 6.87 (d, J = 2.1 Hz)J=8.9 Hz, H-5'), and 7.59 (dd, J= 2.2, 8.9, Hz, H-6'); 3-glucosyl were δ 5.58 (d, J= 7.7 Hz, H-1), 3.64 (dd, J= 7.7, 9.3 Hz, H-2), 3.53 (t, J=9.4 Hz, H-3), 3.34 (t, J=9.3 Hz, H-4), 3.33 (ddd, J= 1.5, 5.6, 9.4 Hz, H-5), 3.81 (*dd*, *J*= 1.5, 11.5 Hz, H-6a), and 3.39 (dd, J= 5.6, 11.5 Hz, H-6b); 2"-rhamnosyl were δ 5.22 (*d*, *J*= 1.5 Hz, H-1), 4.00 (dd, J= 1.5, 3.4 Hz, H-2), 3.79 (dd, J= 3.4, 9.4 Hz, H-3), 3.26 (t_1 J= 9.4 Hz, H-4), 4.07 (dq, J = 6.1, 9.4 Hz, H - 5), and 1.00 $[d, J = 6.1 \text{ Hz}, \text{H-6} (CH_2)]; 6"$ rhamnosyl were δ 4.49 (*d*, *J*= 1.7 Hz, H-1), 3.58 (dd, J = 1.7, 3.4 Hz, H-2), 3.48 (dd, J = 3.4, 9.4 Hz, H-3), 3.22 (t, J = 9.5 Hz, H - 4), 3.41 (dq, J = 6.3)9.5 Hz, H-5), and 1.07 [d, J= 6.3 Hz, H-6 (CH3)]; ¹³C NMR chemical shift (CD₃OD) aglycone were δ 158.94 (C-2), 134.44 (C-3), 179.29 (C-4), 163.15 (C-5), 99.74 (C-6), 165.57 (C-7), 94.69 (C-8), 158.45 (C-9), 105.93 (C-10), 123.45 (C-1'), 117.44 (C-2'), 145.91 (C-3'), 149.54 (C-4'), 116.06 (C-5'), and 123.56 (C-6'); 3-glucosyl were δ 100.50 (C-1), 80.06 (C-2), 78.93 (C-3), 71.89,

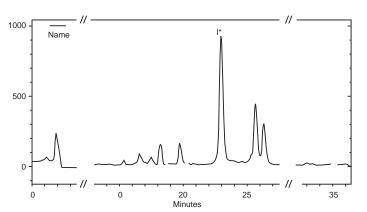


Fig 1–RP-HPLC chromatogram of DMSO/MeOH (1:1) extract of Sesbania javanica Miq. flower. I* is a major compound.

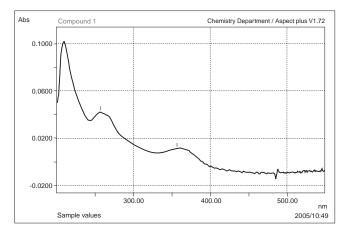
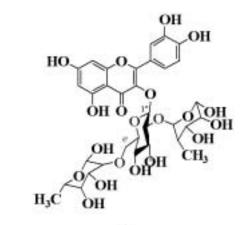


Fig 2–UV-visible spectra of compound I*.



(1)

Fig 3–Structure of the major flavonol glycoside (I*) isolated from the Sesbania javanica Miq. ('Sano') flower, Quercetin 3-2^Grhamnosylrutinoside. 77.10 (C-5), and 68.30 (C-6); 2"-rhamnosyl were δ 102.64 (C-1), 72.41 (C-2), 72.31 (C-3), 74.07 (C-4), 69.96 (C-5), and 17.51 (C-6); and 6"-rhamnosyl were δ 102.26 (C-1), 72.15 (C-2), 72.28 (C-3), 73.88 (C-4), 69.71 (C-5), and 17.81(C-6). Thus, the major compound was identified as quercetin 3-2^G-rhamnosylrutinoside, a known compound. In the quantitation of this compound, 3.16 g/kg of dried weight was calculated from the collection steps.

DISCUSSION

The ¹H NMR spectrum in methanol-d4 of this major compound showed the signals: δ 6.18 (d, J = 2.0 Hz, H-6), 6.38 (d, J = 2.0 Hz, H-8),7.58 (d, J=2.1 Hz, H-2'), 6.87 (d, J=8.9 Hz, H-5'), and 7.59 (dd, J= 2.2, 8.9, Hz, H-6'), and the UV data λ_{max} in MeOH of 357.81 and 256.67 nm for quercetin. The anomeric protons in this major compounds at d 5.58 (d, J= 7.7 Hz, H-1 of glucose), 5.22 (d, J= 1.5 Hz, H-1 of 2"-rhamnose), and 4.49 (d, J= 1.5 Hz, H-1 of 6"-rhamnose) were characteristic of glucose and rhamnose. The configuration of these sugars at C-1 was determined as β for glucose and α for both rhamnoses from the coupling constants. This major flavonol glycoside, quercetin 3-2^Grhamnosylrutinoside, has been reported in the petal of Clitoria ternatea L. and the ¹H NMR and ¹³C NMR spectrum data were identical as those reported by Kazuma et al (2003).

Mutagenic assays, such as the S. typhimurium test, have been widely used to assess the antimutagenic and anticarcinogenic activities of various compounds. The aim of the present study was to evaluate the antimutagenic effects of Sesbania javanica Mig. flower DMSO extract and isolate the major flavonoid against AFB₁ and B(a)P using the Ames test. The results of this study demonstrate that Sesbania javanica Miq. flower DMSO extract and quercetin 3-2^Grhamnosylrutinoside are capable of inhibiting the mutagenicity of AFB1 and B(a)P. These results indicate that the Sesbania javanica Mig. flower contains antimutagens inhibiting the mutagenicities of AFB1 and B(a)P. These results concur with the study of Nakahara et al (2002) with 80% methanol extract.

Most chemical carcinogens, such as AFB₁, B(a)P and PAHs, require metabolic activation to DNA-reactive intermediates by cytochrome P450-mediated mixed -function oxidases (MFOs), undergoings epoxidation of their aromatic ring, in order to exert their carcinogenic action (Dipple et al, 1984). These epoxide intermediates as an electrophilic metabolites covalent binding to cellular DNA, leading to adduct formation, are considered to be critical to the initiation of carcinogenesis. The mechanism by which the antimutagens in Sesbania javanica Miq. flower DMSO extract and the major isolated flavonol glycoside inhibited the mutagenicities of AFB₁ and B(a)P are not known. The constituents in the DMSO crude extract and quercetin 3-2^G-rhamnosylrutinoside may interact with some certain enzymes to inhibit activity which necessary for the activation of these chemical mutagens in the liver (cytochrome P450 enzymes). Many studies have shown that quercetin has antimutagenic effects against AFB1 and B(a)P (Buening et al, 1981; Huang et al, 1983; Ogawa et al, 1985; Francis et al, 1989; Choi et al, 1994). Many papers have reportes the activity of flavonoids as inhibitors of enzymes involved in biotransformation of precarcinogens. These compounds are now seen as potential chemopreventive agents (Elangovan et al, 1994; Mooni et al, 1998; Zhai et al, 1998; Ciolino and Yeh, 1999).

A variety of naturally occurring compounds have been shown to inhibit the mutagenicity or carcinogenicity. We report the antimutagenic effects of *Sesbania javanica* Miq. flower DMSO extract and its major flavonol glycoside, quercetin 3-2^G-rhamnosylrutinoside against AFB₁ and B(a)P mutagens. This is an edible vegetable flower commonly found in Thai food.

ACKNOWLEDGEMENTS

The authors would like to thank Mr Paul Adams for his critical reading this manuscript.

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